

## **DESCRIPTION**

### **"NEW THERAPEUTIC USE FOR A GROUP OF SULPHATED POLYSACCHARIDES"**

#### **Technical field of the invention**

The present invention relates to the use of a sulphated polysaccharide for the preparation of a medicament for the treatment or prophylaxis of arthrosis. Likewise, the present invention relates to the use of a sulphated oligosaccharide for the preparation of a medicament for the treatment or prophylaxis of arthrosis.

#### **Description of the prior art**

Arthrosis, also known as osteoarthritis, is a degenerative joint disease which affects most people over 65 years of age, characterised by a gradual degradation of the cartilaginous tissue, together with the presence of inflammation and pain. Inflammation appears especially when the disease is in an advanced stage and is different in nature from the inflammation observed in rheumatoid arthritis, and generally is only a minority component of arthrosic disease. Arthrosis may be defined as the degeneration of the hyaline articular cartilage. A secondary effect thereto is affectation of the sinovial membrane and the subchondral bone, as well as the formation of new bone on the margins of the joint surfaces. The etiology of arthrosis is unknown and its evolution is slow.

The cartilage allows bones to move, slipping over one another. It also absorbs the stress produced by physical movement. In arthrosis, the surface of the cartilage breaks and wears out, causing bones to move against one another, which leads to friction, pain, swelling, and loss of joint movement. As time goes on, the joint may deform.

As an aid in the diagnosis of arthrosis, two techniques are mainly used: X-rays, which is the simplest method to identify the bone changes in the joint, and nuclear magnetic resonance (NMR), which, unlike the former, makes it possible to simultaneously view all the joint's components.

Under normal conditions, cartilage renewal is a very slow process, which consists of a constant synthesis (anabolism) and degradation (catabolism) of the extracellular matrix components. The chondrocyte is the cell responsible for this metabolism, which must be coordinated.

Under pathological conditions, this process is altered, because cartilage

renewal is accelerated, which leads to a precocious repair of the cartilaginous tissue caused by an imbalance between the chondrocyte's anabolic and catabolic programmes, which entails degradation of the cartilage. The repair reaction is the result of a hyperproliferation of chondrocytes, jointly with an increase in the synthesis of the cartilage's extracellular matrix components by these cells (D. Hamermam *et al.*, *N. Engl. J. Med.*, 320, 1322-1330 (1989)). Consequently, there exists a balance between synthesis and degradation of the cartilage which controls said homeostatic reaction and which depends on systemic hormones and growth factors whose secretions decrease with age. Cartilage degradation is regulated by enzymes and free radicals produced by adjacent tissues, but also by the chondrocyte itself.

We will highlight the following current pharmacological treatments for arthrosis:

Symptomatic-action substances which have a rapid action, such as analgesics, non steroidal anti-inflammatory drugs (NSAIDs), corticoids, and cyclo-oxygenase 2 inhibitors (COX-2).

Symptomatic-action substances which act in a slower manner, known as SYSADOA (Symptomatic Slow Acting Drug for Osteoarthritis) (M.G. Lequesne, *Rev. Rhum. (Eng./Ed.)*, 61, 69-73 (1994)), which include hyaluronic acid, chondroitin sulphate, and glucosamine sulphate. This group is characterised in that the effect starts after 2-3 weeks of treatment and persist for 2 to 6 months after administration ceases.

Documents have been found in the literature about other applications for the sulphated polysaccharides of the present invention:

R.V. Jones (US 2,686,779) discloses a procedure to prepare inulin sulphate salts with alkaline metals. In said patent it is stressed that inulin sulphate alkaline salts with various degrees of sulphation have been applied in the chemical industry as paste thickeners, adhesives, and mud additives used in oil well drilling.

It has been described that inulin sulphate exhibits anti-coagulant (*Arkiv for kemi, mineralogi o. geologi.*, Bd 24B (5), 1-4 (1946)) and antilipemic activity (*Arch. Int. Pharmacodyn.*, XCIX, 334 (1954)).

V.G. Nair *et al.* (US 4,021,545) reveal that inulin polysulphate salts (completely sulphated) have a complement-inhibiting activity, such that, according to the inventors, they could be used in the treatment of diseases such as rheumatoid arthritis, systemic erythematous lupus, certain types of vasculitis, etc.

G. Franz *et al.* (*Bioactive Carbohydrate Polymers*, 47-58, Kluwer Academic

Publishers, Netherlands, 2000) describe pullulan sulphate's anti-coagulant and anti-thrombotic activity.

C.C. Lee (EP 499.164) claims, albeit without describing it in any example, an injectable composition which may contain carrageenan in order to supplement natural lubricating fluids. In the cited document they do not mention carrageenan polysulphate.

T. Kanamaru *et al.* (US 5,135,920) disclose that curdlan sulphate is an angiostatic agent. S. Alban *et al.* (*Thromb. Res.* 78 (3), 201-10 (1995)) describe that the same compound acts as an anti-coagulant and anti-thrombotic agent.

T. Hata *et al.* (JP 04257509) show the application of alginic acid sulphate as a moisturising agent in cosmetic preparations.. L. Lange *et al.* (ES 2.064.736) describe the use of alginate sulphate to inhibit pancreatic cholesterol.

E. Besterman (*Brit. Med. J.*, 310 (1957, I)) describes that the polysaccharide laminarin sulphate exhibits antilipemic activity. H. Q. Miao *et al.* (*Int. J. Cancer*, 83 (3), 424-31 (1999)) reveal laminarin sulphate's anti-metastatic activity. P. Bohlen *et al.* (WO 03/006645) describe that the polysaccharide laminarin sulphate inhibits heparanase activity, which, as they claim, is associated, among others, to inflammatory processes, diabetes, or arthritis.

L. Lange *et al.* (ES 2.064.736) reveal that the polysaccharide pectin sulphate is useful for treating high blood cholesterol levels.

Until now no description has been found of the use of inulin sulphate, gellan sulphate, pullulan sulphate, curdlan sulphate, alginic acid sulphate, laminarin sulphate, or pectin sulphate with any degree of sulphation, or of carrageenan polysulphate in the treatment or prophylaxis of arthrosis.

We also have not found thus far any description of the use of the sulphated oligosaccharides corresponding to the above-mentioned sulphated polysaccharides in the treatment or prophylaxis of arthrosis.

From all of the foregoing we may conclude that providing a useful drug for the treatment of arthrosis is still a therapeutic problem.

#### Disclosure of the invention

The problem to be solved by this invention is to provide alternative sulphated polysaccharides to be used in the treatment or prophylaxis of arthrosis. The solution corresponding to the first aspect of the invention refers to the use of a sulphated polysaccharide in acid form or as a physiologically acceptable salt thereof, selected from the group consisting of inulin sulphate, gellan sulphate, pullulan sulphate,

curdian sulphate, alginic acid sulphate, laminarin sulphate, and pectin sulphate, for the preparation of a medicament for the treatment or prophylaxis of arthrosis in a mammal.

In a preferred embodiment, the sulphated polysaccharide is selected from the group consisting of inulin sulphate, gellan sulphate, pullulan sulphate, and curdian sulphate. Among these, the most preferred are inulin sulphate and gellan sulphate.

In a more preferred embodiment, the sulphated polysaccharide is totally or partially salified with an alkaline or alkaline earth metal. Preferably, the alkaline metal is sodium and the preferred polysaccharide is inulin sulphate sodium salt. The polysaccharide gellan sulphate sodium salt is equally preferred.

Inulin sulphate sodium salt with a degree of sulphation between 25% and 62%, on anhydrous base, is more preferred.

Inulin sulphate sodium salt with a degree of sulphation between 55% and 62%, on anhydrous base, is even more preferred.

In another equally preferred embodiment, the sulphated polysaccharide is partially hydrolysed.

In another equally preferred embodiment, the sulphated polysaccharide is a polysulphated polysaccharide, selected from the group consisting of inulin polysulphate, gellan polysulphate, pullulan polysulphate, curdian polysulphate, alginic acid polysulphate, laminarin polysulphate, and pectin polysulphate. Among these, the most preferred one is inulin polysulphate, preferably inulin polysulphate sodium salt. Equally preferred is gellan polysulphate, preferably gellan polysulphate sodium salt.

Another aspect of this invention is the use of carrageenan polysulphate for the preparation of a medicament for the treatment or prophylaxis of arthrosis in a mammal.

Another aspect of this invention is the use of a sulphated oligosaccharide in acid form or as a physiologically acceptable salt thereof, derived from a polysaccharide selected from the group consisting of inulin, gellan, pullulan, curdian, alginic acid, laminarin, and pectin, for the preparation of a medicament for the treatment or prophylaxis of arthrosis in a mammal. Preferably, the sulphated oligosaccharide is derived from inulin. Also preferably, the sulphated oligosaccharide is derived from gellan.

In a preferred embodiment, the sulphated oligosaccharide is a polysulphated oligosaccharide.

Another aspect of this invention is the use of a sulphated oligosaccharide in acid form or as a physiologically acceptable salt thereof, produced by chemical synthesis, whose structure corresponds to a portion of the structure of a sulphated polysaccharide selected from the group consisting of inulin sulphate, gellan sulphate, pullulan sulphate, curdlan sulphate, alginic acid sulphate, laminarin sulphate, and pectin sulphate, for the preparation of a medicament for the treatment or prophylaxis of arthrosis in a mammal.

In an especially preferred embodiment, the medicament is adapted for oral administration.

Likewise, in an especially preferred embodiment, the drug is adapted for intra-articular administration.

The sulphated polysaccharides in this invention are well-known, and may be produced by partial or total sulphonation of the free hydroxyl groups present in the polysaccharides' structure: inulin, gellan, pullulan, curdlan, alginic acid, laminarin, and pectin. Carrageenan polysulphate is produced by total sulphonation of the free hydroxyls present in carrageenan.

Sulphonation of a hydroxyl group gives rise to the corresponding sulphate group.

Carrageenan already contains some sulphate groups within its structure. In this invention, carrageenan polysulphate is used.

When we speak herein about the degree of sulphation, this refers to the % of sulphate groups on an anhydrous base (with respect to the molecule).

When in the specification of the present invention we speak about sulphated polysaccharides, this relates to a sulphated polysaccharide with any degree of sulphation. In the literature, a sulphated polysaccharide with a high degree of sulphation is usually called a polysulphated polysaccharide and, naturally, when the polysaccharide is totally sulphated it is also called a polysulphated polysaccharide. Consequently, the present invention includes polysulphated polysaccharides (inulin polysulphate, carrageenan polysulphate, gellan polysulphate, pullulan polysulphate, curdlan polysulphate, alginic acid polysulphate, laminarin polysulphate, and pectin polysulphate).

Sulphated polysaccharides may be produced by partial or total sulphonation of the free hydroxyls in the above-mentioned commercial polysaccharides, by means of procedures described in the literature, such as, for example, by means of chlorosulphonic acid/pyridine (T. Astrup *et al.*, *Acta Physiol. Scand.*, 8, 215-226 (1944)), or else by means of chlorosulphonic acid/dimethylformamide (P.Vongchan,

*Carbohydrate Research*, 337, 1239-1242 (2002)), or else using piperidine-N-sulphonic acid (N. Kinzo *et al.*, *Carbohyd. Res.*, 21, 420-426 (1972)), or else using the sulphur trioxide-pyridine complex (C. Mähner *et al.*, *Carbohyd. Res.*, 331, 203-208 (2001)).

In order to produce various degrees of sulphation, the above-mentioned procedures may be modified (changing the reagent concentrations, the reaction temperature, reaction times, etc...). Also, if so desired, different polysaccharide hydroxyls may be selectively sulphonated, by means of protection/deprotection of the polysaccharide hydroxyls.

If the starting polysaccharide's average molecular weight is too high, it can, if so desired, be partially hydrolysed before or after sulphonation.

The sulphated oligosaccharides used in the invention may be produced by enzymatic or chemical hydrolysis (procedures which are well-established in the literature) of the commercial polysaccharide, followed by sulphonation, or else directly by enzymatic or chemical hydrolysis of the sulphated polysaccharide. They may also be produced by chemical synthesis.

When we speak in the specification of the present invention about a physiologically acceptable salt, this refers to a salt of an alkaline metal, such as for example sodium or potassium, a salt of an alkaline-earth metal, such as for instance calcium or magnesium, or else an organic salt, such as for instance salt with trimethylamine, with triethylamine, or with an amino acid, such as for instance lysine, arginine, proline, glycine, or serine.

Salification of the sulphated polysaccharide may be performed by simple, well-known chemical procedures.

For use in the treatment or prevention of arthrosis, the sulphated polysaccharides of the invention, whether in acid form or as a physiologically acceptable salt thereof, are formulated in appropriate pharmaceutical compositions, using conventional techniques and excipients or carriers, such as the ones described in *Remington's Pharmaceutical Sciences Handboock*, Mack Pub. Co., N.Y., USA.

The pharmaceutical compositions of the invention may be administered to the patient in required doses. Administration of the compositions may be performed by various means, for example, oral, intravenous, intra-peritoneal, intra-articular, subcutaneous, intramuscular, topical, sublingual, intradermal, or intranasal. The pharmaceutical compositions of the invention include a therapeutically effective quantity of the sulphated polysaccharide in this invention, in acid form or as a

physiologically acceptable salt thereof, with said quantity being dependent on many factors, such as for example, the patient's physical condition, age, sex, specific compound, means of administration, and other factors well-known in technology. Furthermore, it is understood that said dosage of the active compound may be administered in single- or multiple-dose units in order to provide the desired therapeutic effects. If so desired, other therapeutic agents may be used jointly with the ones provided by this invention.

In general, the pharmaceutical preparations in the invention will be in solid or liquid form, or in the form of a gel. The solid-form pharmaceutical preparations which may be prepared in accordance with this invention include powders, minigranules (pellets), pills, dispersible granules, capsules, suppositories, and other solid galenical forms. The liquid-form preparations include solutions, suspensions, emulsions, and microspheres. Preparations in solid form which, immediately before being used, may be converted to liquid preparations for oral, parenteral, or intra-articular administration, are also contemplated. Said liquid forms include solutions, suspensions, and emulsions.

#### Brief description of the figure

Figure 1 represents a Western Blot. The upper arrow indicates the area of intact aggrecan and the lower arrow indicates the area of aggrecan fragments labelled with antibody NITGE G1. In the first of the wells a control sample was seeded (without any treatment); in the second one, a sample treated with interleukin-1 (IL-1 $\alpha$ ); in the third, a sample treated with 100  $\mu$ g/mL of inulin polysulphate in Chemical Example 1, and, in the fourth, a sample treated with IL-1 $\alpha$  and 100  $\mu$ g/mL of inulin polysulphate in Chemical Example 1.

#### Detailed description of the invention

The following examples are merely illustrative and do not represent a limitation on the scope of this invention.

##### Chemical Example

##### Example 1: Preparation of inulin polysulphate sodium salt

##### 1-A: Preparation of inulin polysulphate sodium salt using chlorosulphonic acid/pyridine

Under constant stirring, at a temperature < 6°C, 88 mL (1.32 mol; 1.8 eq /OH) of chlorosulphonic acid were added drop by drop to 580 mL (7.20 mol) of dry pyridine. The resulting mixture was heated to 75°C and, subsequently, 40 g (0.25

mol) of inulin (Fibruline Standar, marketed by Trades S.A.) were added. The stirring and the heating to 100°C were continued for 5 h. Once this time elapsed, the reaction mixture was allowed to cool down to approximately 50°C and 50 mL of de-ionised water were added in order to destroy the chlorosulphonic acid excess. During this step, the ambient temperature was slightly increased and two phases appeared.

The crude reaction product was allowed to rest until it reached ambient temperature. The upper phase (pyridine) was separated by suction, while the oily lower phase was poured over a 10 % solution of sodium acetate in methanol. The precipitate that was formed was allowed to sediment. The supernatant was separated by decanting and discarded.

Treatment with the sodium acetate methanolic solution was repeated twice and, in the last step, the solid was separated by vacuum filtering by means of a type k-100 depth filter (Pall Corporation. Seitz-k-100).

The filter produced (brown solid) was dissolved in distilled water and the resulting solution was vacuum filtered by means of a k-100 depth filter in order to eliminate the insoluble matter remainders resulting from the reaction.

The filtrate resulting from the preceding step was a dark amber colour solution which was treated with quaternary ammonium (Quartamin). As a consequence, a very abundant solid was formed which was isolated by vacuum filtering using a Hyflo earth pre-layer (100 g). The filter was abundantly washed with distilled water

The inulin polysulphate-Quartamin complex was carefully separated from the filtering earths and treated with a 20 % NaCl aqueous solution, at 80°C, for two hours.

Once the two hours of reaction elapsed, the heating was stopped, and when the ambient temperature reached 60°C, isopropanol was added. Stirring was maintained for 30 min. and, subsequently, the mixture was poured into a decanting funnel in order to separate the aqueous phase from the organic one.

The organic phase (isopropanol + quaternary ammonium) was discarded. The aqueous phase was vacuum filtered in order to eliminate the Hyflo earth remainders.

A mixture of methanol / acetone was poured over the aqueous phase. The brown precipitate that was formed was allowed to sediment. The supernatant was separated by decanting and discarded. The sediment was washed with methanol and subsequently dissolved in de-ionised water.



The pH of the resulting solution was adjusted between 10.5-11 with 10% NaOH. The solution was heated to 50°C and treated with H<sub>2</sub>O<sub>2</sub> for 15 min. Finally, the heating was stopped and the pH of the medium was adjusted to 5.5 with a 2% acetic acid solution.

When the discoloured solution reached ambient temperature, a 10% methanol / sodium acetate solution was poured thereon. A white or slightly yellow precipitate immediately formed which was allowed to sediment.

The supernatant was separated by decanting and the sediment was abundantly anhydrated with methanol.

The inulin sulphate was separated by vacuum filtering with a No. 3 porous plate and dried in a vacuum heater at 30°C until the methanol concentration was equal to or lower than 0.3 %.

The product was produced in the form of a fine, amorphous powder, white or slightly yellow in colour.

Determination of the degree of sulphation (organic sulphates) in the produced inulin polysulphate sodium salt

Exactly 150 mg of the product were weighed, they were dissolved in water, and the resulting solution was taken to 250 mL with the same solvent. 5 mL of this solution were pipetted, they were transferred to a 50 mL beaker, and 25 mL of water were added. It was photometrically assessed at 420 nm, with a 0.1% (0.00279 M) N-cetylpyridinium chloride (CPC) solution.

The sulphate content was calculated using the following equation:

$$\%SO_4 = \frac{V \times 0.00279 \times 96 \times 100}{5 \times (100 - PPS) \times W} = \frac{13.392 \times V \times 100}{100 - PPS} \times W$$

250 × 100                      100

Where:

V: Volume of the 0.1% CPC solution consumed, in mL.

W: Weight of the sample, in mg.

PPS: Weight loss by desiccation of the product (105°C, 4 h), in %.

Degree of sulphation produced: 58.0%, expressed on anhydrous base, which corresponds to a completely sulphated inulin molecule (sulphonation of all the hydroxyls)

Free residual pyridine and pyridine in salt form: 127 ppm (the permitted limit is 200 ppm)

Free sulphates: 0.08%

IR (KBr) cm<sup>-1</sup>: 2900 (C-O-H), 1247 (S=O), 800 (C-S-O)

1-B: Preparation of inulin polysulphate sodium salt using chlorosulphonic acid/dimethylformamide (DMF)

In a three-neck reactor (2L), having a thermometer and two addition funnels, 1,000 mL of dry DMF were added under a nitrogen current. The solvent was cooled to a temperature of  $< 6^{\circ}\text{C}$ , and 112 mL (1.68 mol; 1.8 eq /OH) of chlorosulphonic acid were added drop by drop under constant stirring.

The reactor temperature was allowed to reach  $15^{\circ}\text{C}$ , and, subsequently, a solution of 52.6 g of inulin in 400 mL of dry DMF was added. The reaction was maintained at ambient temperature for less than 3 h.

Once this time elapsed, the reaction was poured over cold water. The pH of the resulting solution was increased to 7-8 with 30 % NaOH and, subsequently, the solution was poured over a 10 % solution of sodium acetate in methanol. The solid that precipitated was allowed to sediment and was separated from the supernatant.

The solid was separated by vacuum filtering through a Büchner funnel. It was dissolved in 1,500 mL of process water and the solution was treated with 350 mL of quaternary ammonium (Quartamin). The white solid that was formed was isolated through vacuum filtering using a Hyflo earth pre-layer (100 g).

The solid retained in the filter, jointly with the Hyflo earth pre-layer, were transferred to a reactor endowed with coolant and mechanical stirring, to which 3,000 mL of a 20 % NaCl solution were added. The mixture was stirred at  $80^{\circ}\text{C}$  for 6 h and it was subsequently vacuum- and heat-filtered, using filter paper. The filtrate (A) was saved, and the solid retained in the Büchner funnel was transferred once again to the reactor in order to repeat the complex breakdown step under the same exact conditions. The filtrate (B) which was produced from this repetition was mixed with (A), and the residual solid was discarded.

3.3 volumes of a methanol / acetone mixture were poured over the solution produced from the union of filtrates (A) and (B). The solid which precipitated was allowed to sediment and was isolated by suction of the supernatant. The sediment was washed with methanol, vacuum-drained through a Büchner funnel, and dissolved in process water. The NaCl content in this solution was determined and adjusted to 10% (w/V).

The precipitation step using methanol / acetone was repeated, and the solid deposited on the bottom of the contained was dissolved in de-ionised water.

The sodium chloride concentration in this solution was adjusted to 2 % (w/V) and, subsequently, 2 volumes of methanol were added. The mixture was allowed to rest. The product was deposited on the bottom of the container. The supernatant

was separated by suction and the product was dissolved in process water in order to obtain a solution with an approximate concentration of 5 % (w/V). This solution was diafiltered through a regenerated cellulose membrane MWCO=5,000-1,000 Da. The permeate generated during this step was periodically assayed with  $\text{AgNO}_3$ . The diafiltering was considered to be finished when the chloride assay gave a negative result.

The diafiltered solution was lyophilised. The lyophilate that was produced was ground and sieved. The inulin polysulphate sodium salt was produced in the form of an amorphous powder, white or slightly yellow in colour.

#### Pharmacological examples

The cartilage's resistance and repair capacity are determined by the proteoglycans of the extracellular matrix, particularly the aggrecans. Synthesis of these aggrecans by the articular chondrocytes, and their quality, diminishes with age, which is one of the main factors involved in the development of arthrosis.

#### Example 1: Evaluation of the *in vitro* effects of inulin polysulphate on the synthesis of extracellular aggrecans in a primary culture of human chondrocytes

This procedure may be applied to the evaluation of any sulphated polysaccharide of this invention.

##### 1A: Aggrecan levels determined by the incorporation of $^{35}\text{S}$

##### Materials and methods

Human articular chondrocytes were isolated following the methods described by W.T. Green Jr. (*Clin. Orthop.*, 75, 248-260 (1971)) and K.E. Kuettner *et al.* (*J. Cell. Biol.*, 93, 743-750 (1982)).

These chondrocytes were cultured in an agarose gel following the method described by P.D. Benya *et al.* (*Cell*, 30, 215-224, (1982)), and modified by G. Verbruggen *et al.* (*Clin. Exp. Rheumatol.*, 8, 371-378 (1990)) and by M. Cornelissen *et al.* (*J. Tiss. Cult. Meth.*, 15, 139-146 (1993)).

The synthesis of aggrecans was determined by the incorporation of  $^{35}\text{S}$ , using labelled sodium sulphate  $\text{Na}_2^{35}\text{SO}_4$  as radioactive precursor. After two weeks of culture, 10  $\mu\text{Ci/mL}$  of radio-labelled precursor were introduced in the culture medium for 48 h, as was the compound to be assayed (inulin polysulphate in Chemical Example 1) at different concentrations (0.0, 0.1, 0.5, 1.0, 5.0  $\mu\text{g/mL}$ ).

The synthesised  $^{35}\text{S}$  aggrecans once again partially accumulated in the intercellular agarose matrix or else were released in the incubation medium.

Once the incubation period ended, the agarose gel was mechanically broken down and subsequently digested by means of 3 mL of a 50 U/mL agarose solution

in a 0.067 M phosphate buffer, at pH 6.0, in the presence of proteinase inhibitors.

The suspension thus produced was centrifuged; the supernatant containing the inter-territorial matrix's  $^{35}\text{S}$  aggrecans and the incubation medium containing the  $^{35}\text{S}$  aggrecan metabolites released in the extracellular matrix were subsequently united by means of chromatography.

The residue, which contained the chondrocytes and the associated  $^{35}\text{S}$  aggrecans, was treated for 48 hours with 1 mL of a 4.0 M guanidium chloride solution in a 0.05M acetate buffer at pH 5.8 containing the proteinase inhibitors.

The purpose of this operation is to extract the  $^{35}\text{S}$  aggrecans associated with the cells. The solution produced was centrifuged in order to separate the cells from the supernatant, which was subsequently separated by means of chromatography. The chromatography operations for the different fractions produced were performed on Sephadex G25 gel in a phosphate buffer pH 6.8 containing 0.01 M of  $\text{Na}_2\text{SO}_4$ , in order to separate the  $^{35}\text{S}$  aggrecans from the free  $\text{Na}_2^{35}\text{SO}_4$ .

The radioactivity of each of the macromolecular eluants produced was measured and related to the number of chondrocytes contained in the initial culture, expressed in pg of  $^{35}\text{SO}_4$  incorporated into the aggrecans, per million chondrocytes per hour.

### Results

They are shown in Tables 1, 2, and 3.

The tables show the remarkable effectiveness of the inulin polysulphate in Chemical Example 1, in the total production of re-synthesised aggrecans (Table 3). The inulin polysulphate in Chemical Example 1 is also capable of increasing production of aggrecans in the inter-territorial matrix (Table 1) as well as of the aggrecans associated with the cells (Table 2).

The optimum activity is found at a concentration of about 0.5  $\mu\text{g/mL}$ .

Table 1

pg $^{35}\text{S}/10^6$ cells/h	% change	% Total	IPS ( $\mu\text{g/mL}$ )
7,710		67.4	0.0
7,818	1.4	66.2	0.1
12,383	60.6	70.7	0.5
10,887*	41.2*	70.7	1.0
8,340	8.2	67.9	5.0

Quantity of radioactively-labelled sulphur incorporated into the inter-territorial matrix's aggrecans, per million cells per hour. % Change= Percentage change with respect to the untreated control (IPS= 0.0  $\mu\text{g/mL}$ ). % Total= Percentage with respect to the total quantity of aggrecans. IPS: dose of inulin polysulphate in Chemical Example 1 added. \*  $p < 0.05$  with respect to the non-treated control.

Table 2

pg $^{35}\text{S}/10^6$ cells/h	% change	% Total	IPS ( $\mu\text{g/mL}$ )
2,378		20.9	0.0
2,737	15.1	23.5	0.1
3,064*	28.8*	18.4	0.5
2,662	11.9	17.9	1.0
2,041	14.2	17.4	5.0

Quantity of radioactively-labelled sulphur incorporated into the aggrecans of the inter-territorial matrix associated with the cells, per million cells per hour. % Change= Percentage change with respect to the non-treated control (IPS= 0.0  $\mu\text{g/mL}$ ). % Total= Percentage with respect to the total quantity of aggrecans. IPS: dose of inulin polysulphate in Chemical Example 1 added. \*  $p < 0.05$  with respect to the non-treated control.

Table 3

pg $^{35}\text{S}/10^6$ cells/h	% change	% Total	IPS ( $\mu\text{g/mL}$ )
11,429		100	0.0
11,756	2.9	100	0.1
17,312*	51.5*	100	0.5
15,281*	33.7*	100	1.0
12,106	3.0	100	5.0

Quantity of radioactively-labelled sulphur incorporated into the re-synthesised aggrecans, per million cells per hour. % Change= Percentage change with respect to the non-treated control (IPS= 0.0  $\mu\text{g/mL}$ ). % Total= Percentage with respect to the total quantity of aggrecans. IPS: dose of inulin polysulphate in Chemical Example 1 added. \*  $p < 0.05$  with respect to the non-treated control.

### 1B: Aggrecan levels determined by means of immunocytochemistry

#### Materials and methods

The human articular chondrocytes were isolated and cultured following the previously described methods.

The accumulation of aggrecans associated with the chondrocytes was measured, after a week of culture, by means of the immunocytochemistry technique described by L. Wang *et al.* (*Osteoarthritis Cart.*, 9, 248-260 (2001)). The chondrocytes were treated with interleukin-1 (IL-1) in order to simulate a situation of inflammation and catabolism. The inulin polysulphate in Chemical Example 1 was added jointly with IL-1.

Once the incubation period ended, the chondrocytes were released from the agarose gel following the described method. The chondrocytes and the associated aggrecans were studied by means of flux cytometry using monoclonal antibodies specifically directed against the aggrecans' proteic part.

#### Results

They are shown in Table 4.

This Table confirms the effectiveness of the inulin polysulphate in Chemical Example 1 on the production of aggrecans associated with the cells.

Moreover, it can be seen that there is a dose-effect relationship, since, with greater doses of the compound, there is a greater increase in the production of aggrecans associated with the cells.

Table 4

	Donor M32		Donor F45	
	M±SD	%Change	M±SD	%Change
Control	18.9±0.2	105.6	36.5±0.2	201.7
IL-1	17.9±0.3	100.0	18.1±0.2	100.0
IL-1+IPS (0.5 µg/mL)	-----	-----	20.4±0.2*	112.7*
IL-1+IPS (1.0 µg/mL)	20.8±1.2*	116.2*	26.6±0.1*	147.0*
IL-1+IPS (2.5 µg/mL)	21.6±1.1*	120.7*	39.0±0.2*	215.5*
IL-1+IPS (5.0 µg/mL)	25.0±1.4*	139.7*	-----	-----

Mean fluorescence intensity in the chondrocytes after labelling of aggrecan with antibodies conjugated with fluorescein. M±SD= mean ± standard deviation. % Change= Percentage change with respect to the group treated with IL-1. IL-1 = Interleukin-1. IPS= inulin polysulphate in Chemical Example 1. \* p<0.05 with respect to the group treated with IL-1.

**Example 2:** Determination of inulin polysulphate's capacity to inhibit aggrecan degradation induced by IL-1 $\alpha$

This procedure may be applied to the evaluation of any sulphated polysaccharide of this invention.

#### Materials and methods

The assays were performed on chondrocyte cultures from a rat chondrosarcoma cell line called LTC.

The LTC cells were kept in a monolayer culture in a Gibco DMEM medium (Dulbecco's Modification of Eagle's Medium) (containing 4.5 g/L of glucose) supplemented with sodium bicarbonate (3.7 g/L), glutamine (2mM), ascorbic acid (50 mg/L), gentamicin (50 mg/L), and bovine fetal serum Hyalone (10%), at pH 7.4.

The confluent LTC cell cultures were trypsinised and 40,000 cells per 0.5 mL of culture were seeded in 48-well plates. They were maintained for 5 days, during which time they deposited 15-30 µg of glycosaminoglycans (GAG) in each well.

The medium was separated, the cell membranes were washed by adding 5 x 1 mL of a catabolic medium {DMEM + 4.5 g/L glucose supplemented with sodium bicarbonate (3.7 g/L), glutamine (2 mM), and 10 ng/mL of IL-1 $\alpha$  (interleukin-1 $\alpha$ ) at pH 7.4}. Subsequently, the cultures were kept in 200 µL of this medium, supplemented or not with inulin polysulphate sodium salt in Chemical Example 1, for

4 days, without changing the medium, at 37 °C.

In order to verify inulin polysulphate sodium salt's effect on aggrecan degradation, a Western Blot was performed.

For the Western Blot, 20 µL of 50 mM tris-acetate at pH 7.3 were added to each well. The cultures (medium + cell membrane) were deglycosylated with 50 mU of chondroitinase ABC at 37 °C for 4 hours. Subsequently, the samples in each well were recovered and centrifuged at 1,460 g for 10 minutes. 25 µL-fractions of the supernatant were separated by electrophoresis ( gel gradient 8-12 % SDS PAGE) and analysed by means of a Western Blot using anti-aggrecan antibody G1. G1 is the product of the aggrecanase activity, also verified by immunoreactivity with the anti-NITGE neo-epitope-specific antibody. The antiserum was used at a 1:5,000 dilution and the peroxidase-conjugated goat anti-(rabbit IgG) was detected by means of the Amersham ECL Kit.

### Results

They are shown in Figure 1, where a Western Blot is represented which is the result of the various treatments applied to a rat chondrosarcoma chondrocyte culture. In said figure, it may be seen that treatment of the chondrocytes with IL-1 $\alpha$  causes an increase in aggrecan degradation, such that the band of aggrecan fragments labelled with antibody NITGE G1 becomes thicker. On the contrary, when the culture is treated with inulin polysulphate in Chemical Example 1, in the presence or not of IL-1 $\alpha$ , aggrecan degradation is inhibited, such that the above-mentioned band narrows down until it almost disappears. Consequently, inulin polysulphate is capable of reducing the aggrecan degradation in a chondrocyte culture, for which reason it will also stop the degradation of the extracellular matrix in the case of patients with arthrosis.